

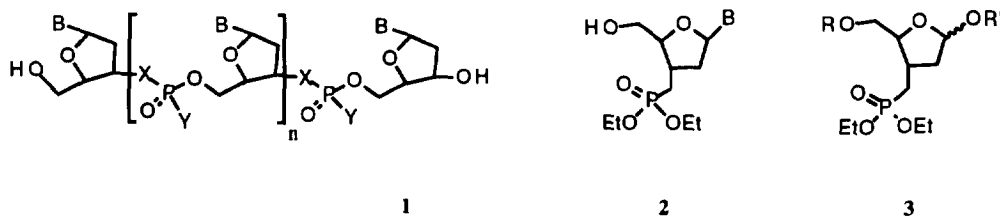


PREPARATION OF 3'-PHOSPHONATE ANALOGS OF 2',3'-DIDEOXYNUCLEOSIDES

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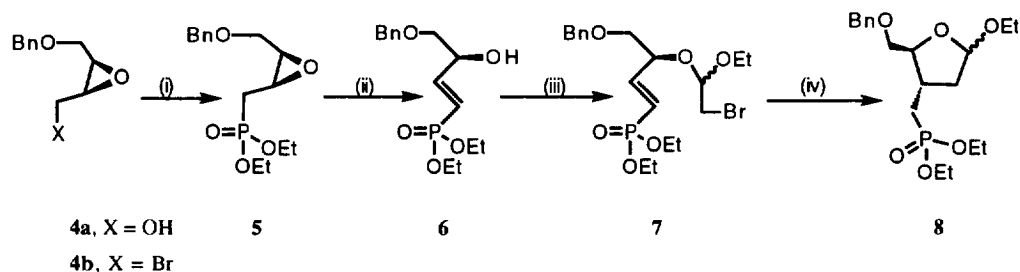
Summary: γ -Hydroxyl vinylphosphonate (6) has been efficiently prepared by the base-catalyzed opening of the readily available epoxide (5) and converted to the corresponding nucleosides for antiviral studies. Copyright © 1996 Elsevier Science Ltd

Nucleoside isosteric phosphonates (1) have been recently considered as possible alternatives for oligodeoxynucleosides as antisense agents (Scheme 1).¹ The oligomers (1a, X = CH₂, Y = O⁻) based on phosphonate bond do not introduce new asymmetric centers, and are expected to resist nucleases due to the partial modification of natural phosphate bond.¹ However, antisense properties of the oligomers (1a, X = CH₂, Y = O⁻) have not been studied owing to a lack of readily available starting materials.² Phosphonates (2) can serve as useful building blocks for the preparation of compounds (1a). The potential value of nucleoside phosphonates (2) has prompted us to design a synthesis for a key intermediate such as phosphonate (3), which could lead to either the corresponding nucleotide oligomers (1)³ or the nucleoside analogs (2) for antiviral studies.⁴ Herein, we describe a strategy for the construction of optically active compounds (2) via the phosphonate intermediate (3).⁵



Scheme 1

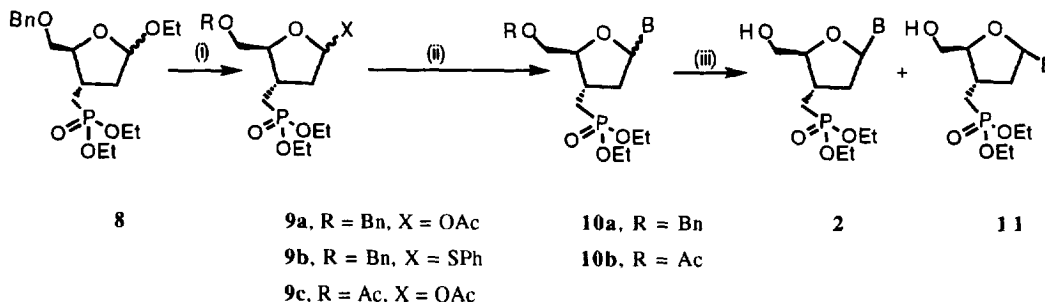
Central to this synthetic strategy are base-catalyzed opening of a chiral epoxide and furan ring formation via the radical cyclization⁶ of a vinylphosphonate (Scheme 2). Compound (**4a**) was prepared according to a literature procedure⁷ and was converted to the corresponding bromide (**4b**) in 89% yield. Although the Arbusov reaction of (**4b**) with trimethyl phosphite was not successful due to the rapid conversion of phosphite to dimethyl methylphosphonate, the use of neat triethyl phosphite produced the desired product (**5**) in 95% yield. The reaction of 10 grams of β,γ -epoxyphosphonate (**5**) with a catalytic amount of sodium ethoxide (0.1 eq.) in 100 ml of ethanol opened the epoxide within one hour to give the *trans* vinylphosphonate (**6**) in 96% yield. Treatment of (**6**) with the bromoacetal reagent,^{6a} prepared from bromine and ethylvinyl ether, formed bromide (**7**) which upon radical-cyclization afforded (**8**) as a mixture of two diastereomeric products. The stereochemistry of these two products was ascertained to be *trans* 3,4-disubstituted, regardless of the stereochemistry of 1-ethoxyl stereocenter.^{5b,6b} The approach described here is concise and all of the five reactions, including the epoxide-ring opening, the Arbusov condensation, and the Stork cyclization, were also carried out in a large scale resulting in high yields.



Scheme 2: (i) CBr_4 , Ph_3P , MeCN , 0°C to r t, 2 h, 89%; then $\text{P}(\text{OEt})_3$, 140°C , 7 h, 95%. (ii) EtONa (0.1 eq.), EtOH , rt, 1 h, 96%. (iii) Br_2 , ethylvinyl ether, Et_2O , -78°C , then (**6**), Et_3N , rt, 7 h, 74%. (iv) $n\text{-Bu}_3\text{SnH}$, AIBN , PhH , 78°C , 6 h, 88%.

The preparation of nucleoside analogs (**2**) was investigated using different starting materials (**9a-c**) and under various reaction conditions (Scheme 3). The coupling intermediate (**9a**) can be prepared either by PPTS-catalyzed hydrolysis of acetal (**8**), followed by acetylation, or by direct hydrolysis of (**8**) in the presence of acetic acid and anhydride. The other coupling intermediates (**9b**)⁸ and (**9c**) are also easily accessible as shown in Scheme 3. The coupling of (**9a-c**) with trimethylsilylated thymine gave the corresponding phosphonates in good yields, but the α/β selectivity varied between 3:1 to 1:6. Although a continued investigation is required to study the effect of the R group of (**9**) on the resultant stereochemistry of nucleoside-base

coupling, the present method provides both isomers for their antiviral activity studies. The preparation of other analogs containing different nucleoside bases and their anti-HIV activity studies are currently in progress.



Scheme 3. (i) (**8**) to (**9a**): AcOH, Ac₂O, CSA, 70 °C, 0.5 h, 70%; (**9a**) to (**9b**): PhSSiMe₂, ZnI₂, (ClCH₂)₂, rt, 2 h, 74%; (**9a**) to (**9c**): EtOAc, Pd/C, H₂, rt, 0.5 h, 71%, then Ac₂O, Et₃N, DMAP, CH₂Cl₂, rt, 1 h, 73%. (ii) Bis(TMS) thymine, (ClCH₂)₂; (**9a**) to (**10a**): SnCl₄, 2 h, rt, 55%, α:β = 3:1; (**9b**) to (**10a**): NBS, CH₂Cl₂, -78 °C, 1 h, 80%, α:β = 1:2; (**9c**) to (**10b**): BF₃·OEt₂, rt, 2 h, 87%, α:β = 1:6; (iii) For (**10a**), Pd/C, H₂, rt, 1 h, 70%; for (**10b**), K₂CO₃, MeOH, 2 h, 80%.

In conclusion, a very simple and concise approach provides the optically active phosphonate (**8**) in a large quantity which can be converted to phosphonate nucleoside analogs (**2**) and (**11**). Meanwhile, the monomers (**2**) will be used in their insertion into oligodeoxynucleosides to yield substrates such as (**1**), which can be used in testing their resistances to nucleases.

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